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bastien-chan.info

Answers MATH HANDBOOK TRANSPARENCY MASTER Operations with Scientific Notation Addition and Subtraction Before numbers in scientific notation can be added or subtracted, the exponents must be equal. Not equal Equal $(3.4 \times 10^2) + (4.57 \times 10^3) = (0.34 \times 10^3) + (4.57 \times 10^3)$ The decimal is moved to the left to increase the exponent.

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Scientific notation takes the form of $M \times 10^n$ where $1 < M < 10$ and "n" represents the number of decimal places to be moved, Positive n Indicates the standard form is a large number. Negative n Indicates a number between zero and one, Example 1: Convert 1,500,000 to scientific notation,

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SCIENTIFIC NOTATION Name Example 1: Convert 1,500,000 to ...

Math Handbook Transparency Masters Chemistry: Matter and Change 93
Addition and Subtraction Before numbers in scientific notation can be added or subtracted, the exponents must be equal. $(3.4 \times 10^2) + (4.57 \times 10^3)$ # $(0.34 \times 10^3) + (4.57 \times 10^3)$ Not equal Equal The decimal is moved to the left to increase the exponent. # $(0.34 \times 4.57) \times 10^3$ # 4.91×10^3 Multiplication

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94 Chemistry: Matter and Change Math Handbook Transparency Worksheets
1. Perform the following operations and express the answers in scientific notation. a. $(1.2 \times 10^5) + 6(5.35 \times 10^4)$ $(1.2 \times 10^5) - (5.35 \times 10^6)$ $(0.12 \times 10^6) - (5.35 \times 10^6)$ $(0.12 \times 5.35) \times 10^6$ 5.47×10^6 b. $(6.91 \times 10^2) - (2.4 \times 10^3)$

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3) (6.91 10 2) (2.4 10 3) (6.91 10 2) (0.24 10 2)

Scientific Notation Use with Appendix B, Scientific Notation

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1. For each of the measurements in the table below, determine if the underlined number is significant or not significant. Place a check mark in the appropriate box and in the box under the rule that you used to make your determination. 2.

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